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Key indicators

Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.009$ Å
 R factor = 0.075
 wR factor = 0.196
Data-to-parameter ratio = 18.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.3-(4-Bromobenzoyl)-*N*-phenyl-1,3-oxazolidin-2-imine

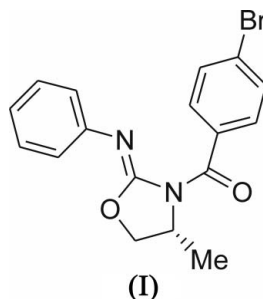
The title compound, $\text{C}_{17}\text{H}_{15}\text{BrN}_2\text{O}_2$, was prepared by the reaction of 2-anilino-4-methyl-2-oxazoline with 4-bromobenzoyl chloride in the presence of potassium *tert*-butoxide at room temperature. X-ray crystallographic analysis shows that the *endo*-substituted product is formed.

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Comment

Alkylation and acylation reactions of 2-amino-2-oxazolines bearing similarly ambident nucleophiles exhibit problems of regioselectivity, occurring either on the endocyclic or exocyclic N atom, depending on the experimental conditions and on the nature of the electrophilic reactants. The reaction of 2-amino-2-thiazolines with some electrophilic compounds has been investigated in detail (Avalos *et al.*, 2000), but 2-amino-2-oxazolines have been less studied (Ganboa *et al.*, 1982; Lee *et al.*, 2002). The reaction of 2-anilino-4-methyl-2-oxazoline with 4-bromobenzoyl chloride furnished regioselectively the *endo*-substituted product 3-(4-bromobenzoyl)-*N*-phenyl-2-oxazolidinimine, (I), and its structure is reported here.



All bond lengths and angles in (I) show normal values (Table 1). The iminoxazoline ring adopts an envelope conformation (Fig. 1), with atom C10 deviating from the N1/C8/C9/O2 plane by 0.149 Å. The N1/C8/C9/O2 mean plane forms dihedral angles of 46.9 (3) and 62.5 (3)° with the C1–C6 and C11–C16 rings, respectively.

Experimental

The title compound was synthesized following the procedure used by Jang *et al.* (2005) for related compounds. To a stirred solution of potassium *tert*-butoxide (0.16 g, 1.43 mmol) and 2-anilino-4-methyl-2-oxazoline (0.27 g, 1.19 mmol) in anhydrous tetrahydrofuran (15 ml) under nitrogen at room temperature, 4-bromobenzoyl chloride (0.34 g, 1.55 mmol) was added dropwise. The solution was stirred for 30 min, then quenched with water (30 ml) and extracted with diethyl ether. The combined extracts were dried over magnesium sulfate, filtered and concentrated. Purification by flash chromatography

afforded the title compound in 71% yield. Single crystals suitable for an X-ray diffraction study were obtained by recrystallization from hexane (m.p. 393–395 K).

Crystal data

$C_{17}H_{15}BrN_2O_2$
 $M_r = 359.22$
 Triclinic, $P\bar{1}$
 $a = 5.867$ (1) Å
 $b = 10.638$ (2) Å
 $c = 13.982$ (3) Å
 $\alpha = 69.44$ (3)°
 $\beta = 78.09$ (3)°
 $\gamma = 86.28$ (3)°
 $V = 799.5$ (3) Å³

$Z = 2$
 $D_x = 1.492$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 34 reflections
 $\theta = 9.6$ – 10.5 °
 $\mu = 2.58$ mm⁻¹
 $T = 298$ (2) K
 Plate, colourless
 $0.30 \times 0.20 \times 0.01$ mm

Data collection

Stoe STADI-4 diffractometer
 $\omega/2\theta$ scans
 Absorption correction: numerical
 (*X-SHAPE*; Stoe & Cie, 1996)
 $T_{\min} = 0.578$, $T_{\max} = 0.895$
 3646 measured reflections
 3646 independent reflections
 2237 reflections with $I > 2\sigma(I)$

$\theta_{\max} = 27.5$ °
 $h = 0 \rightarrow 7$
 $k = -13 \rightarrow 13$
 $l = -17 \rightarrow 18$
 3 standard reflections
 frequency: 60 min
 intensity decay: 3.9%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.075$
 $wR(F^2) = 0.196$
 $S = 1.13$
 3646 reflections
 199 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0712P)^2 + 1.6199P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.98$ e Å⁻³
 $\Delta\rho_{\min} = -0.76$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

Br1—C1	1.898 (5)	N1—C10	1.396 (7)
O1—C7	1.212 (7)	N1—C8	1.473 (7)
O2—C10	1.359 (6)	N2—C10	1.253 (7)
O2—C9	1.455 (7)	N2—C11	1.411 (7)
N1—C7	1.390 (7)		
C10—O2—C9	108.9 (4)	C10—N1—C8	109.1 (4)
C7—N1—C10	129.6 (4)	C10—N2—C11	121.5 (5)
C7—N1—C8	120.2 (4)		

All H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.98 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *STADI4* (Stoe & Cie, 1996); cell refinement: *STADI4*; data reduction: *X-RED* (Stoe & Cie, 1996); program(s)

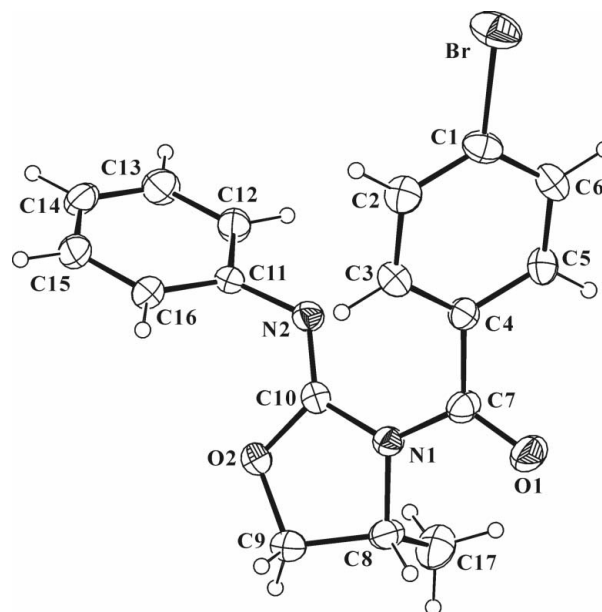


Figure 1

An *ORTEP-3* (Farrugia, 1997) view of (I), with the atom-numbering scheme and 30% probability displacement ellipsoids.

used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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References

- Avalos, M., Babiano, R., Cintas, P., Chavero, M. M., Higes, F. J., Jimenez, J. L., Palacios, J. C. & Silvero, G. (2000). *J. Org. Chem.* **65**, 8882–8892.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Ganboa, I. & Palomo, C. (1982). *Bull. Soc. Chim. Fr.* **2**, 167–170.
 Jang, J. H., Kim, H. J., Kim, J. N. & Kim, T. H. (2005). *Bull. Korean Chem. Soc.* **26**, 1027–1028.
 Lee, G.-J., Kim, T. H., Kim, J. N. & Lee, U. (2002). *Tetrahedron Asymmetry*, **3**, 9–12.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Stoe & Cie (1996). *STADI4*, *X-RED* and *X-SHAPE*. Stoe & Cie GmbH, Darmstadt, Germany.